

# SPH3U Formula Sheet

## Kinematics

$$v_{av} = \frac{\Delta d}{\Delta t}$$

$$\vec{v}_{av} = \frac{\Delta \vec{d}}{\Delta t}$$

$$\vec{a}_{av} = \frac{\Delta \vec{v}}{\Delta t}$$

$$\Delta \vec{d} = \left( \frac{\vec{v}_f + \vec{v}_i}{2} \right) \Delta t$$

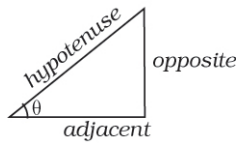
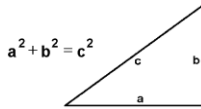
$$\vec{v}_f = \vec{v}_i + \vec{a}_{av} \Delta t$$

$$\Delta \vec{d} = \vec{v}_i \Delta t + \frac{1}{2} \vec{a}_{av} \Delta t^2$$

$$\vec{v}_f^2 = \vec{v}_i^2 + 2\vec{a}_{av} \Delta \vec{d}$$

$$\Delta \vec{d} = \vec{v}_f \Delta t - \frac{1}{2} \vec{a}_{av} \Delta t^2$$

$$\vec{g} = -9.8 \text{ m/s}^2$$



$$\sin \theta = \frac{o}{h}$$

$$\cos \theta = \frac{a}{h}$$

$$\tan \theta = \frac{o}{a}$$

$$x = \frac{-b \pm \sqrt{b^2 - 4ac}}{2a}$$

## Forces

$$\vec{F}_{Net} = m\vec{a}$$

$$\vec{F}_g = m\vec{g}$$

$$\mu_S = \frac{\vec{F}_S}{\vec{F}_N}$$

$$\mu_K = \frac{\vec{F}_K}{\vec{F}_N}$$

## Energy

$$W = \vec{F} \Delta \vec{d}$$

$$W = \vec{F} \cos \theta \Delta \vec{d}$$

$$E_k = \frac{mv^2}{2}$$

$$E_p = mgh$$

$$W_{net} = \frac{mv_2^2}{2} - \frac{mv_1^2}{2}$$

$$W_{net} = \Delta E_k$$

$$E_m = E_p + E_k$$

$$\text{efficiency} = \frac{E_{out}}{E_{in}} \times 100\%$$

$$P = \frac{\Delta E}{\Delta t}$$

$$P = \frac{W_{net}}{\Delta t}$$

$$T_K = T_C + 273$$

$$q = mc\Delta T$$

$$Q_{released} + Q_{absorbed} = 0$$

$$q = mL_x$$

$$A = A_o \left( \frac{1}{2} \right)^{t/h}$$

$$E = mc^2$$

**Table 1** Specific Latent Heats for Various Substances

Substance	Specific latent heat of fusion ( $L_f$ ) (J/kg)	Melting point ( $^{\circ}\text{C}$ )	Specific latent heat of vaporization ( $L_v$ ) (J/kg)	Boiling point ( $^{\circ}\text{C}$ )
aluminum	$6.6 \times 10^5$	2519	$4.0 \times 10^5$	10 900
ethyl alcohol	$1.1 \times 10^5$	-114	$8.6 \times 10^5$	78.3
carbon dioxide	$1.8 \times 10^5$	-78	$5.7 \times 10^5$	-57
gold	$1.1 \times 10^6$	1064	$6.4 \times 10^4$	2 856
lead	$2.5 \times 10^4$	327.5	$8.7 \times 10^5$	1 750
water	$3.4 \times 10^5$	0	$2.3 \times 10^6$	100

**Table 1** Specific Heat Capacities of Common Substances

Substance	Specific heat capacity (J/(kg $\cdot^{\circ}\text{C}$ ))
water	$4.18 \times 10^3$
ethyl alcohol	$2.46 \times 10^3$
ice	$2.1 \times 10^3$
aluminum	$9.2 \times 10^2$
glass	$8.4 \times 10^2$
iron	$4.5 \times 10^2$
copper	$3.8 \times 10^2$
silver	$2.4 \times 10^2$
lead	$1.3 \times 10^2$

## Waves & Sound

$$f = \frac{1}{T} \quad v = \frac{\lambda(m)}{T(s)} \quad v = f\lambda \quad \mu = \frac{m}{L} \quad v = \sqrt{\frac{F_T}{\mu}}$$

$$v = 331.4 \frac{m}{s} + (0.606 \frac{m}{s} / ^\circ C) T \quad M = \frac{\text{air speed of object}}{\text{local speed of sound}} \quad p = \frac{F}{A}$$

$$L_n = \frac{n\lambda}{2} \text{ for } n = 1, 2, 3 \dots \text{ fixed or free ends} \quad L_n = \frac{(2n-1)}{4}\lambda \text{ for } n = 1, 2, 3 \dots \text{ a fixed and a free end}$$

$$f_{obs} = \left( \frac{v_{sound} + v_{detector}}{v_{sound} + v_{source}} \right) f_0 \quad f_1 L_1 = f_2 L_2$$

## Electricity & Magnetism

$$P = \frac{\Delta E}{\Delta t} \quad V = \frac{\Delta E}{Q} \quad I = \frac{Q}{\Delta t} \quad V_{series} = V_1 + V_2 + \dots \quad V_{parallel} = V_1 = V_2 = \dots$$

$$I_{series} = I_1 = I_2 = \dots \quad I_{parallel} = I_1 + I_2 = \dots \quad R = \frac{V}{I} \quad R_{series} = R_1 + R_2 = \dots$$

$$R_{parallel} = \frac{1}{R_1} + \frac{1}{R_2} + \dots \quad \frac{V_p}{V_s} = \frac{N_p}{N_s} \quad \frac{I_s}{I_p} = \frac{N_p}{N_s} \quad \frac{I_s}{I_p} = \frac{V_p}{V_s}$$